SAOS as a predictive tool to assess the residual reactivity in recycled cementitious binders

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Schleibinger Geräte

Building Materials Testing Systems



Cement Chemistry Particle Interaction

Sustainability

Oscillatory

Rheology

Cement Chemistry Particle Interaction

C_3S (\approx cement) hydration:

C₃S grains dissolution and Calcium Silicate Hydrates (C-S-H) precipitation
Attractive electrostatic force development between C-S-H surfaces
C-S-H gel network formation (until cement setting)

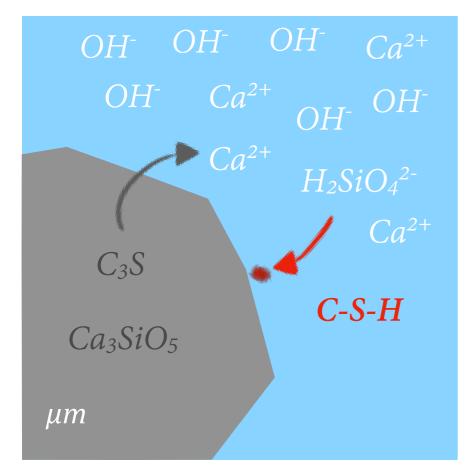
C=CaO, S=SiO₂, H=H₂O

Taylor, H. F., Cement chemistry, 1997.

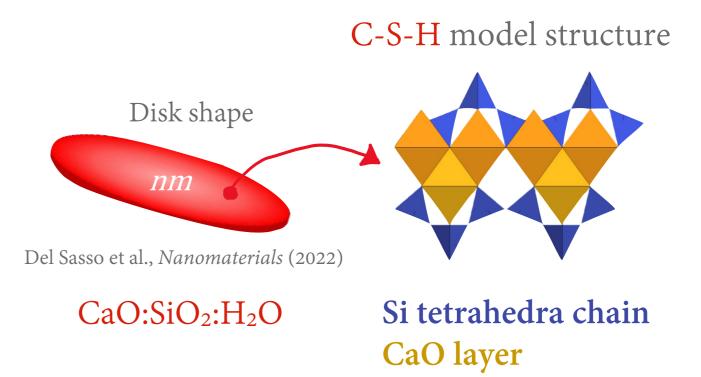


Phase 1: C₃S dissolution & C-S-H precipitation Tricalcium silicate

Calcium Silicate Hydrate

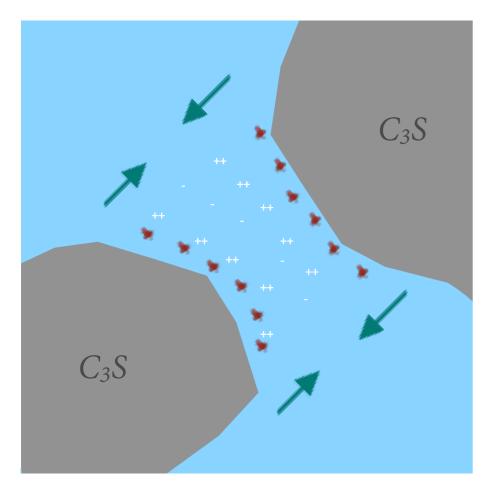


Immediately after contact with water the [OH⁻] increases resulting in a pH>12





Phase 2: Attractive electrostatic force development between C-S-H surfaces \dots and C₃S grains !

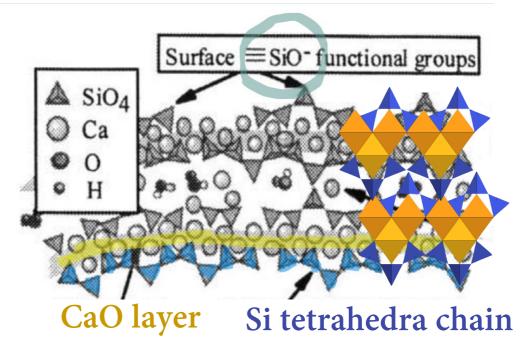


Attractive force & cohesion *f(pore solution concentration)*

Lesko et al., Ultramicroscopy, 2001 Jönsson et al. Langmuir, 2005

C-S-H model structure

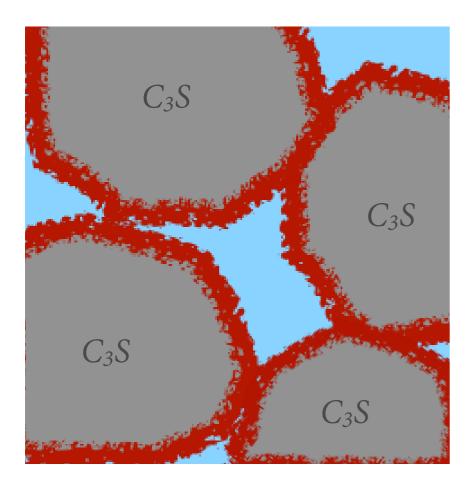
This negatively charged group forms after contact with water reacting with OH⁻ (at high pH)



Iwaida et al. Studies in Surface Science and Catalysis (2001)



Phase 3: C-S-H gel network formation (until cement setting)



C-S-H gel:

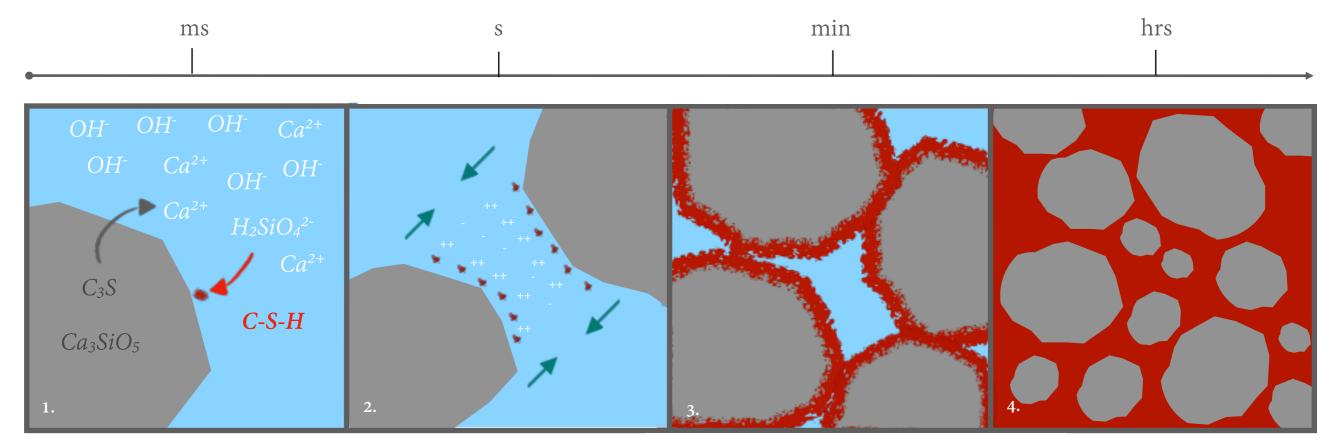
- nano crystalline/nano structurate
- layered structure highly disordered at longer distances (µm)



Liberto et al. *Langmuir* (2022) Del Sasso et al., *Nanomaterials* (2022)



Why does cement set?



1. C₃S grain dissolution and Calcium Silicate Hydrate (C-S-H) precipitation

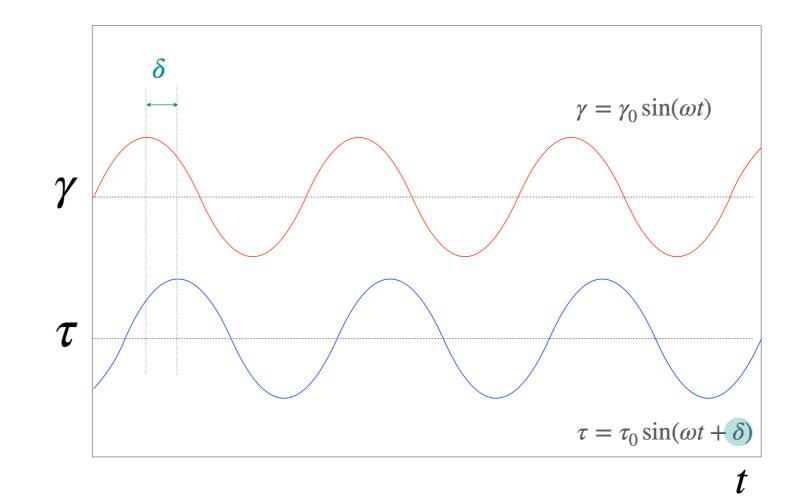
- 2. Attractive electrostatic force development between C-S-H surfaces
- 3. C-S-H gel network formation (until cement setting 4.)

C-S-H the "glue" of cement After complete hydration ca. 80% of the C₃S has been dissolved



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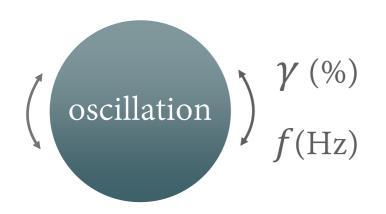
Oscillatory Rheology



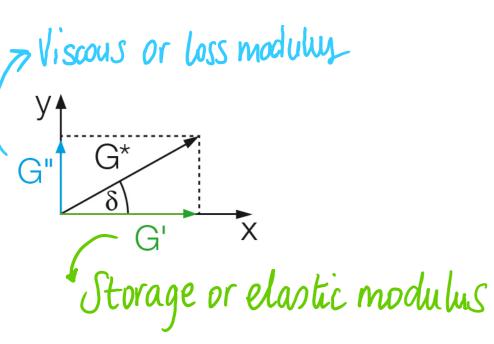
 $\delta = 0$ purely elastic $\delta = \pi/2$ purely viscous $0 < \delta < \pi/2$ viscoelastic

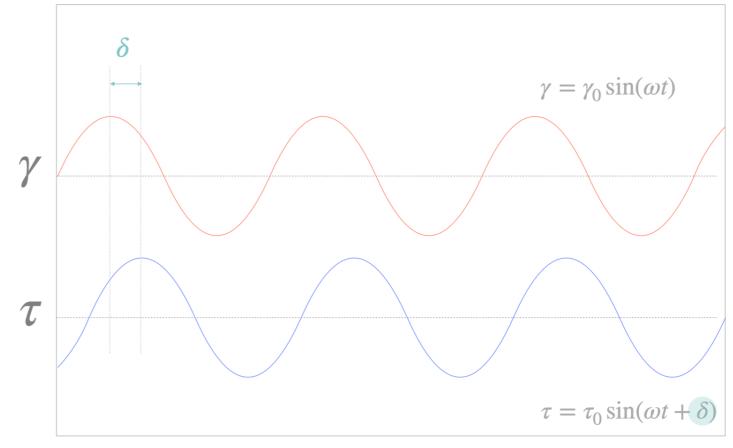






 $G^* = \tau^*(t) / \gamma^*(t)$ $G^* = G' + iG''$

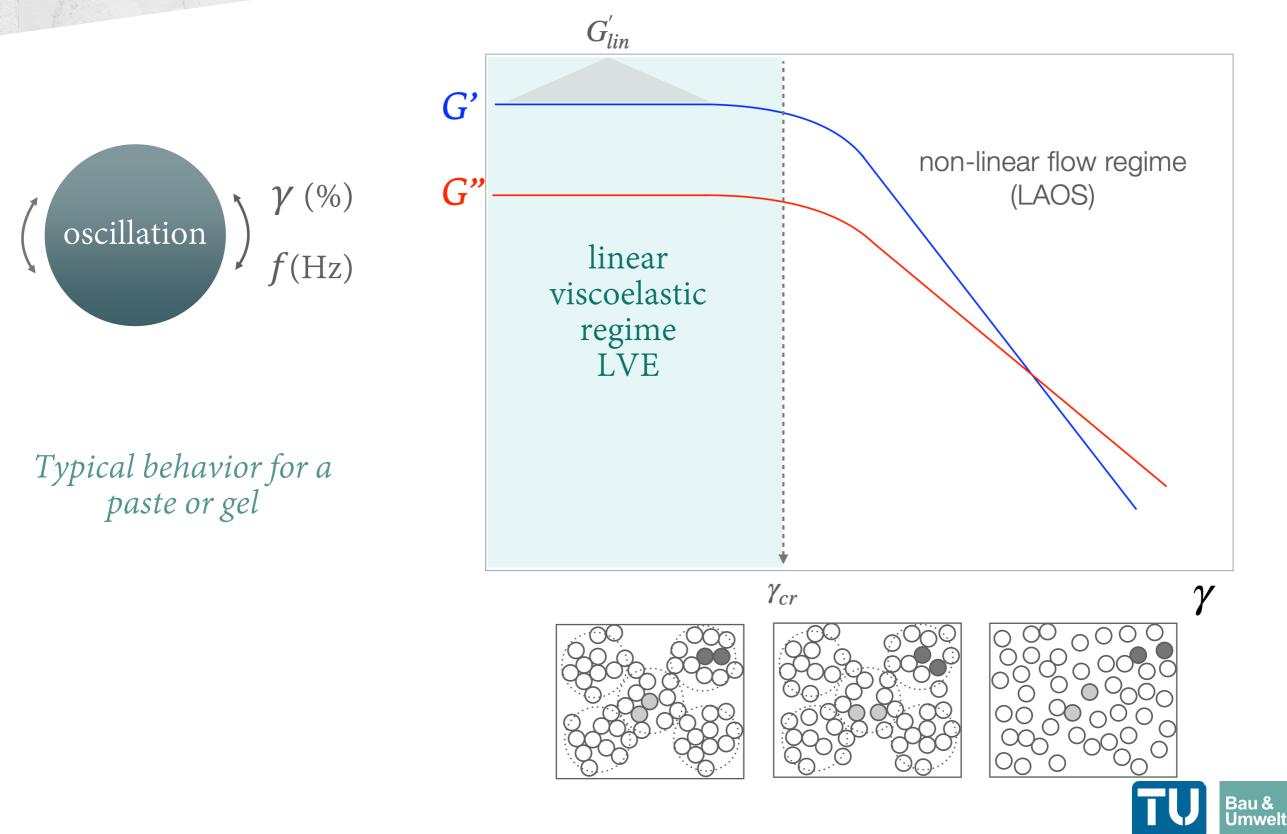




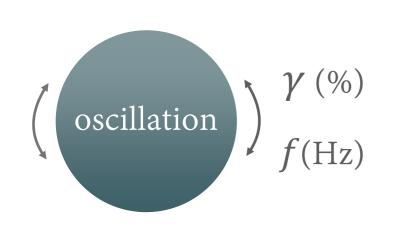
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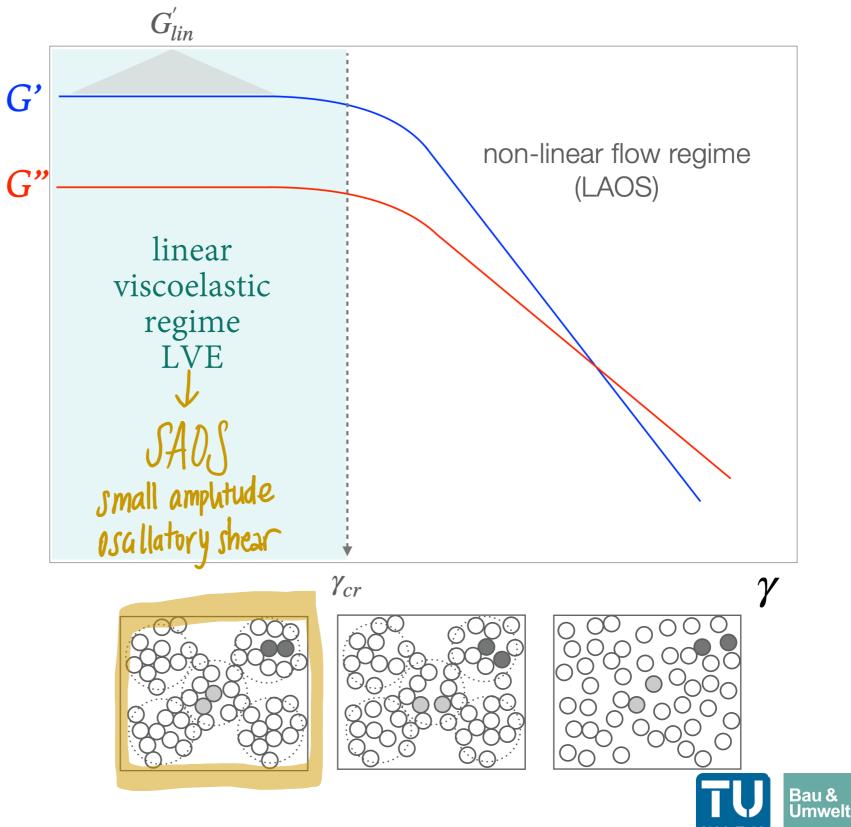
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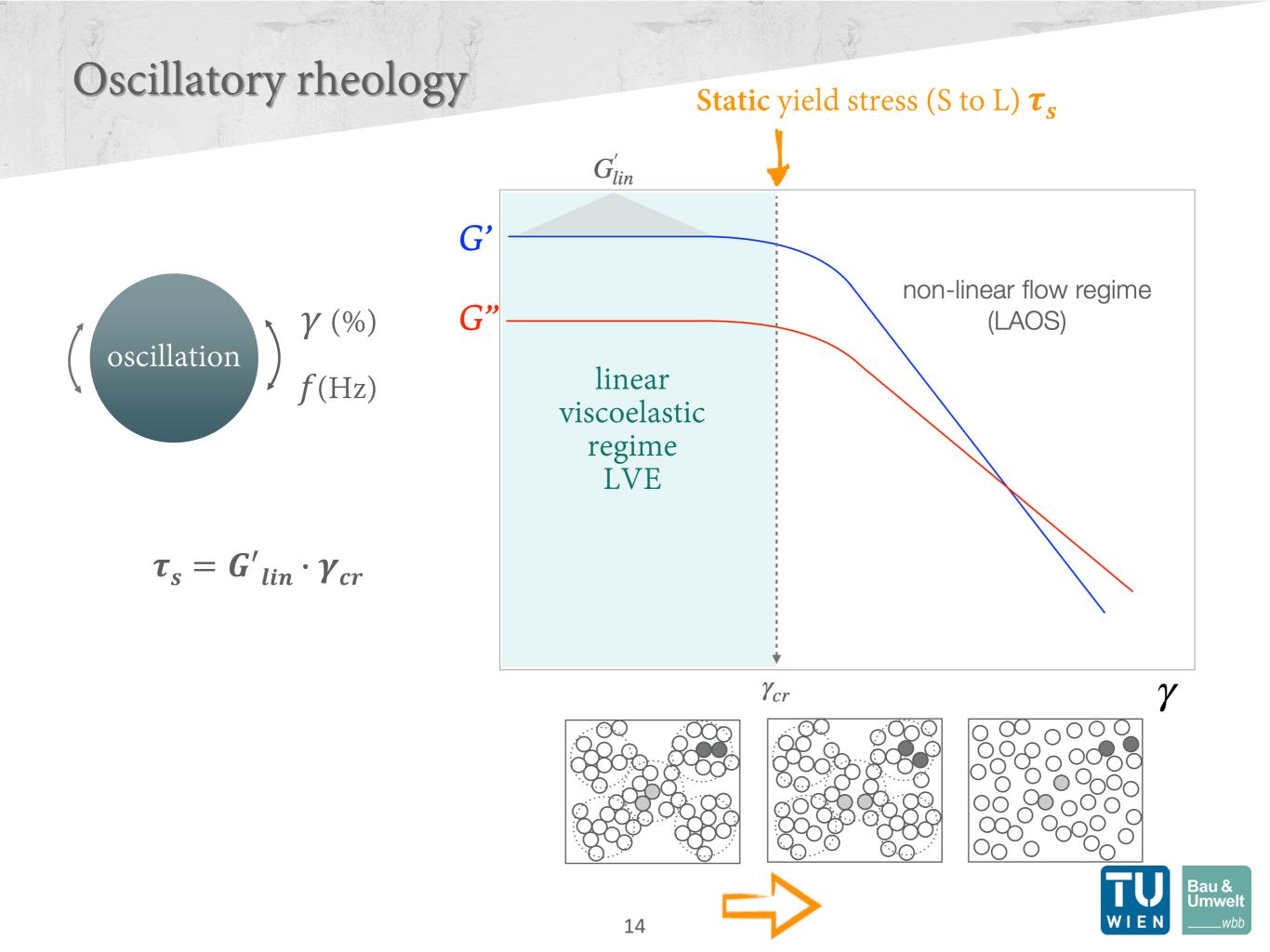


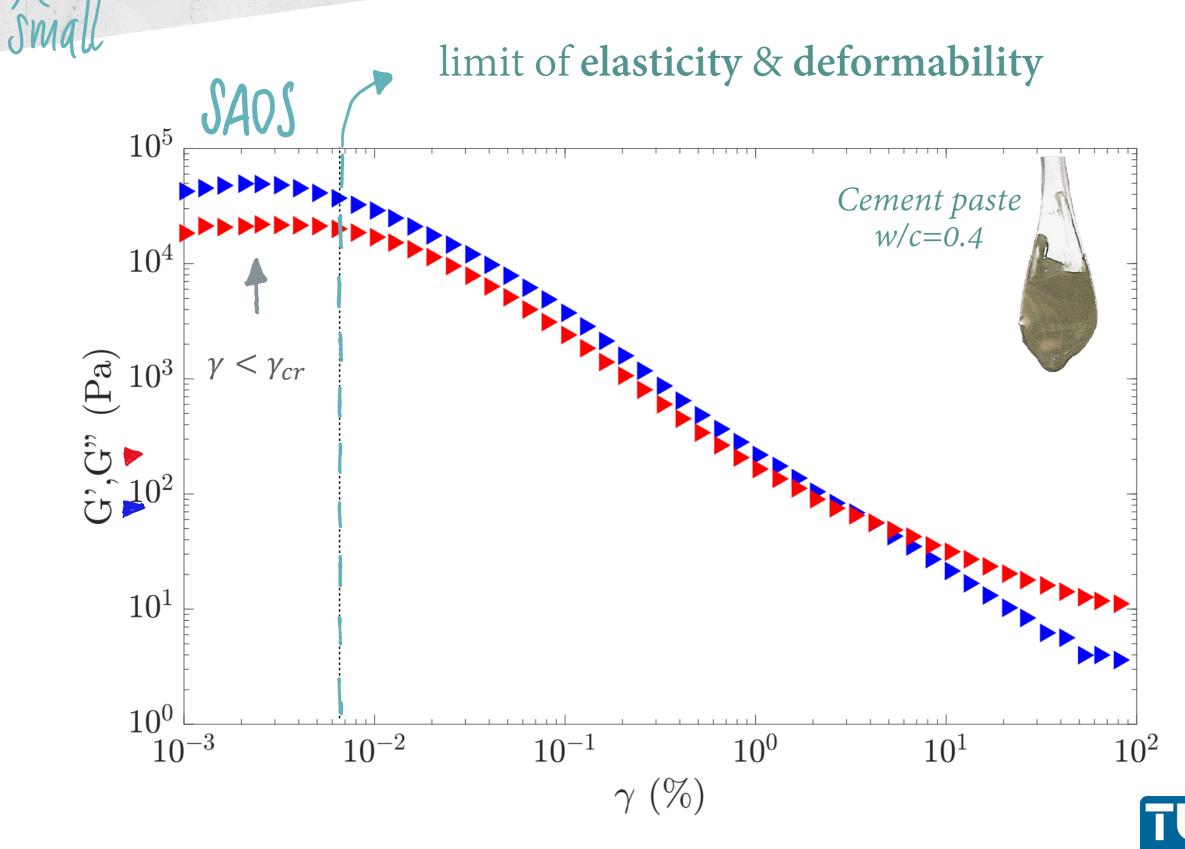
wbb





_wbb

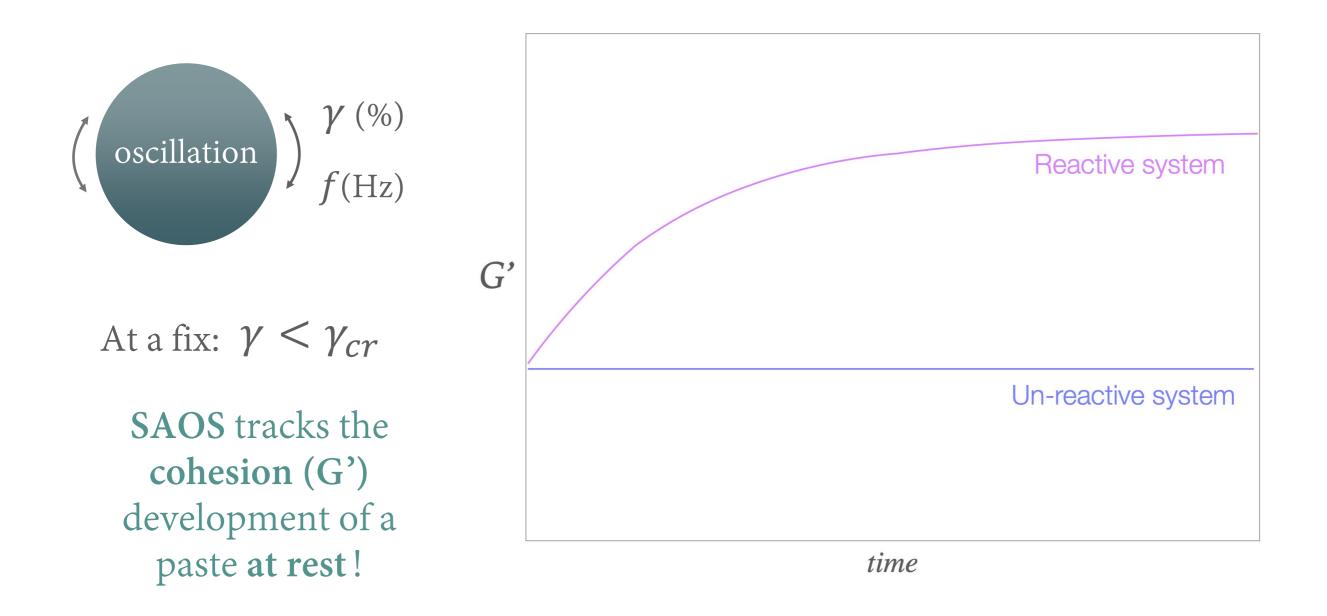




Bau & Umwelt

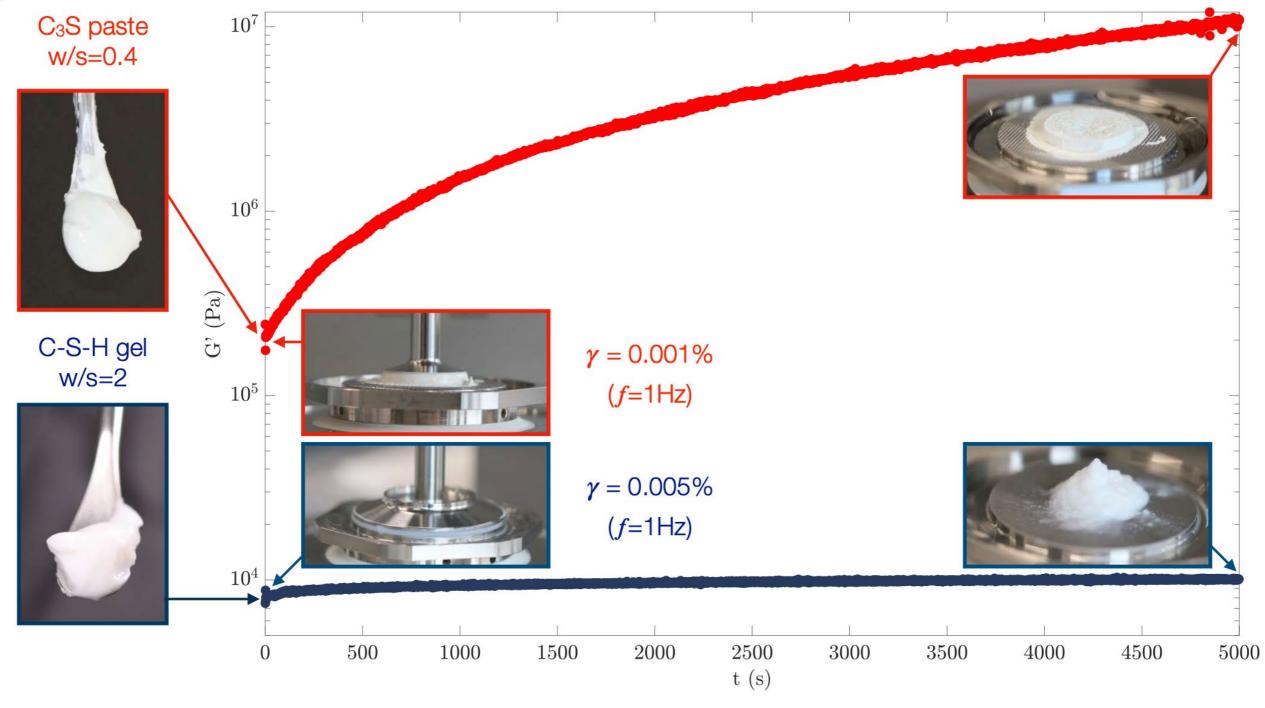
wbb.

small oscillatory rheology SAOS





small oscillatory rheology SAOS



Liberto et al. *Langmuir* (2022)

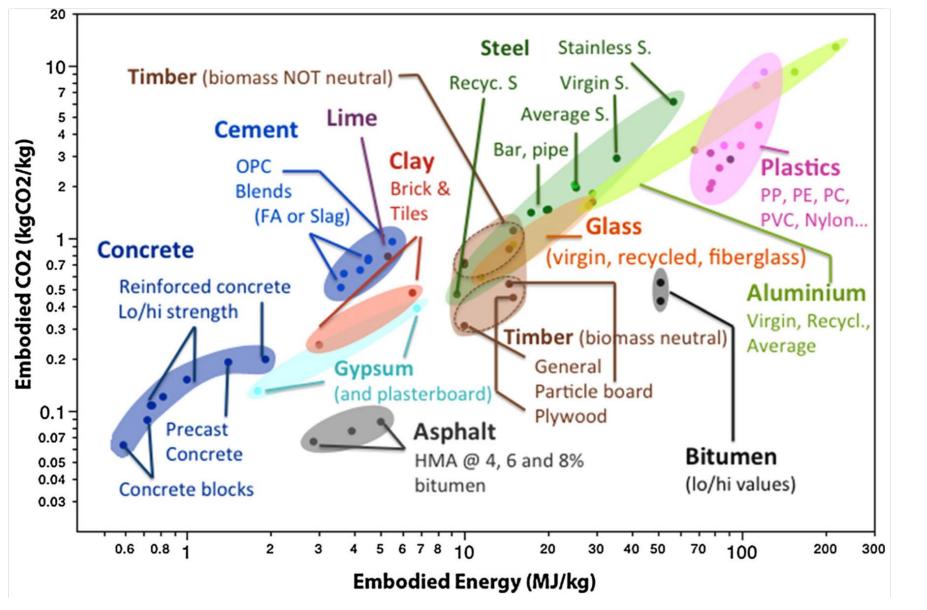


Cement Chemistry Particle Interaction

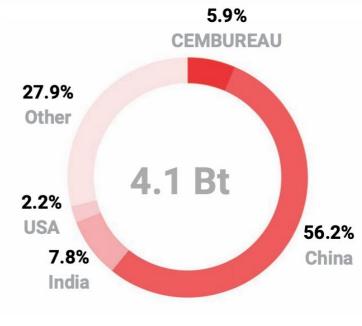
Rheology SAOS

Sustainability

Cement sustainability







Cembureau 2020 report



Barcelo et al. Mater Struct. (2014)

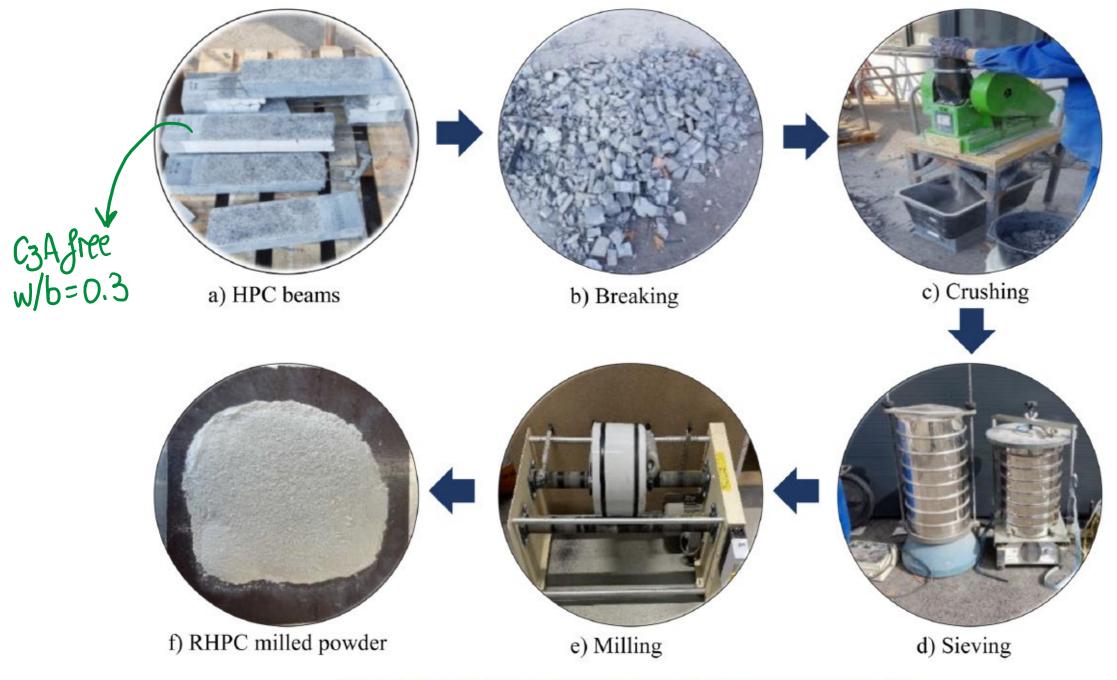


Fig. 1. Recycling process of high-performance concrete (HPC) to obtain mRHPC.



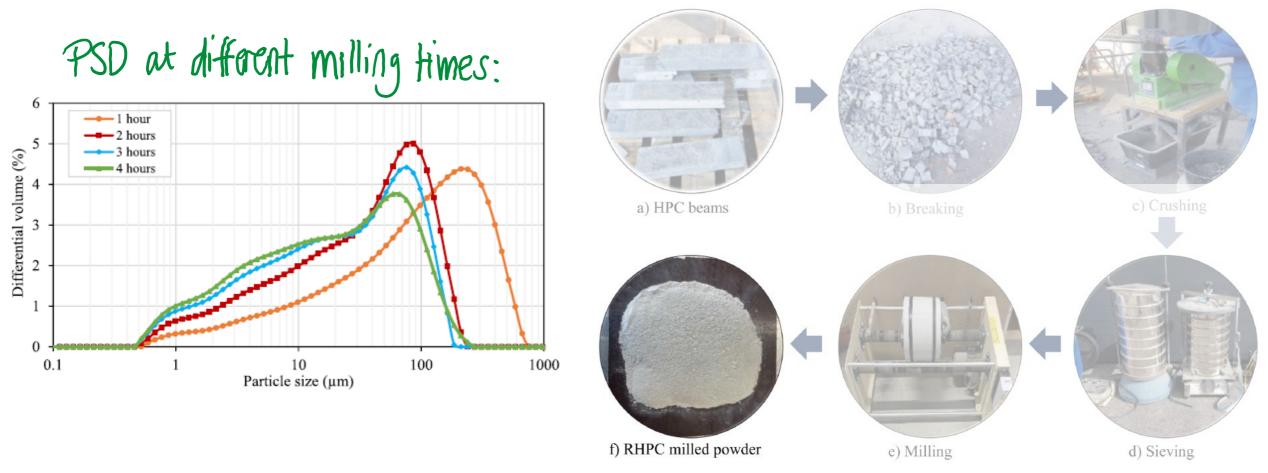


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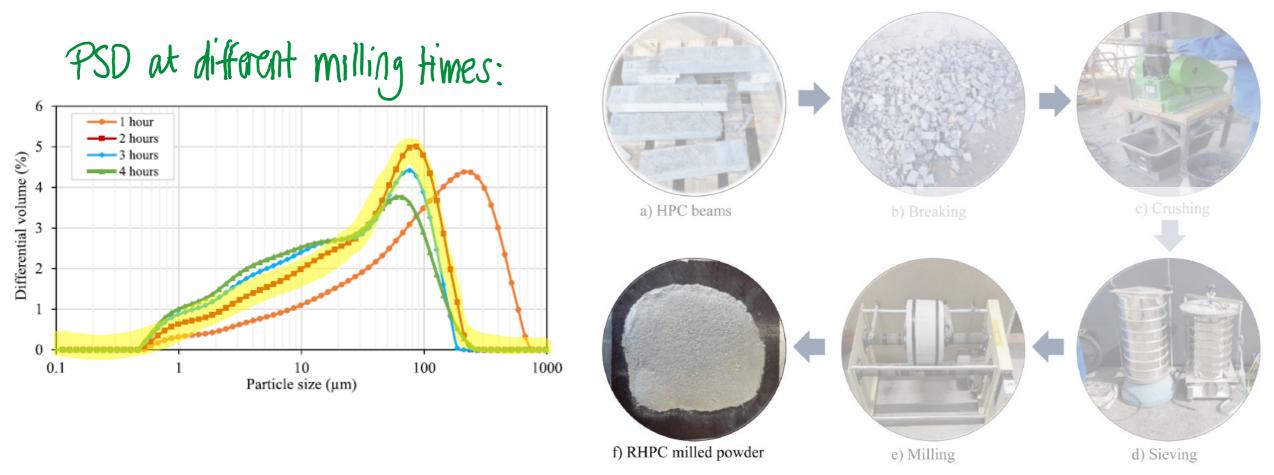
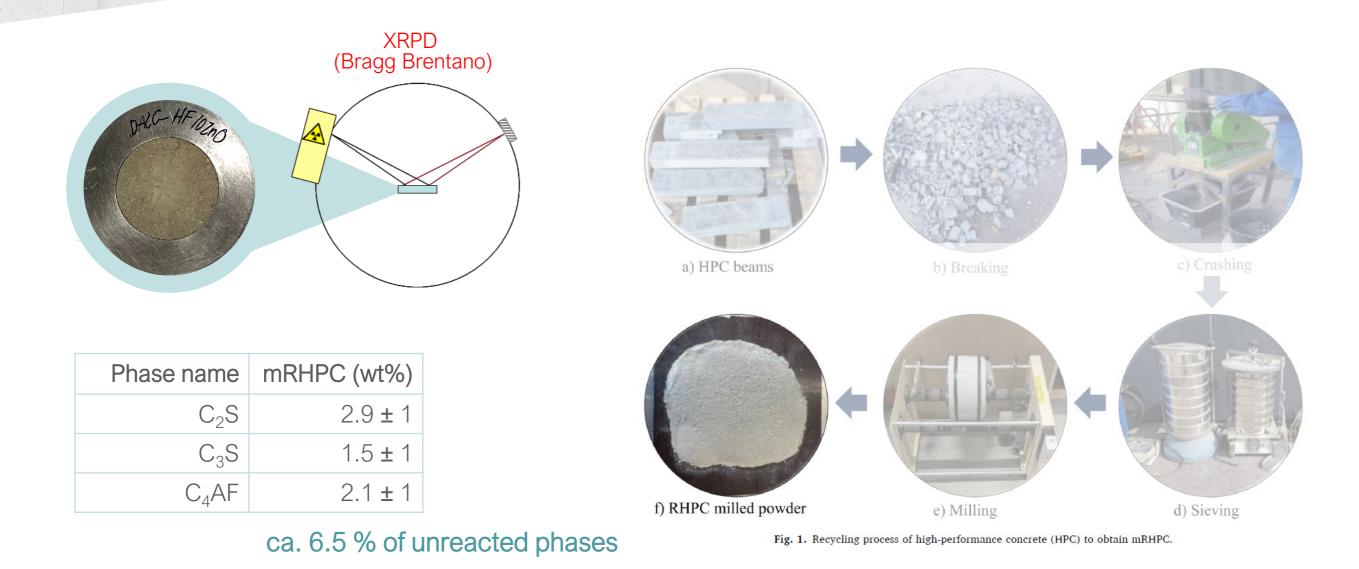
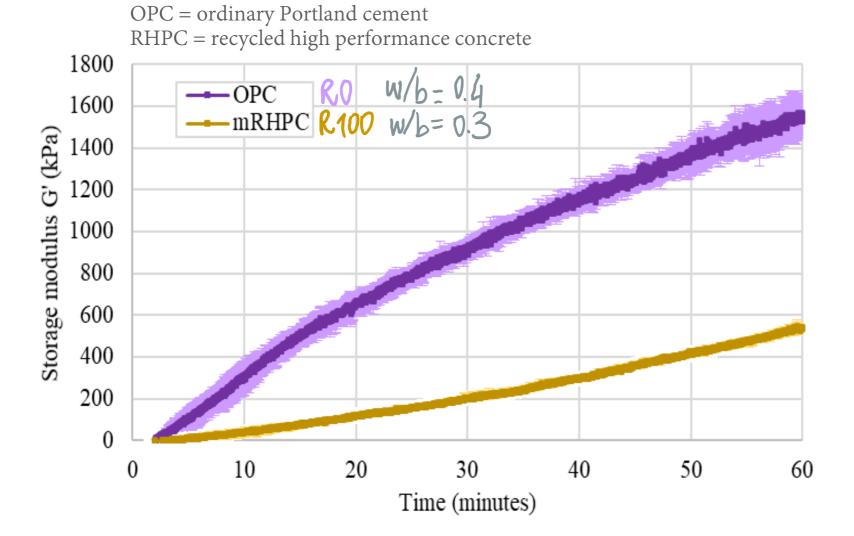


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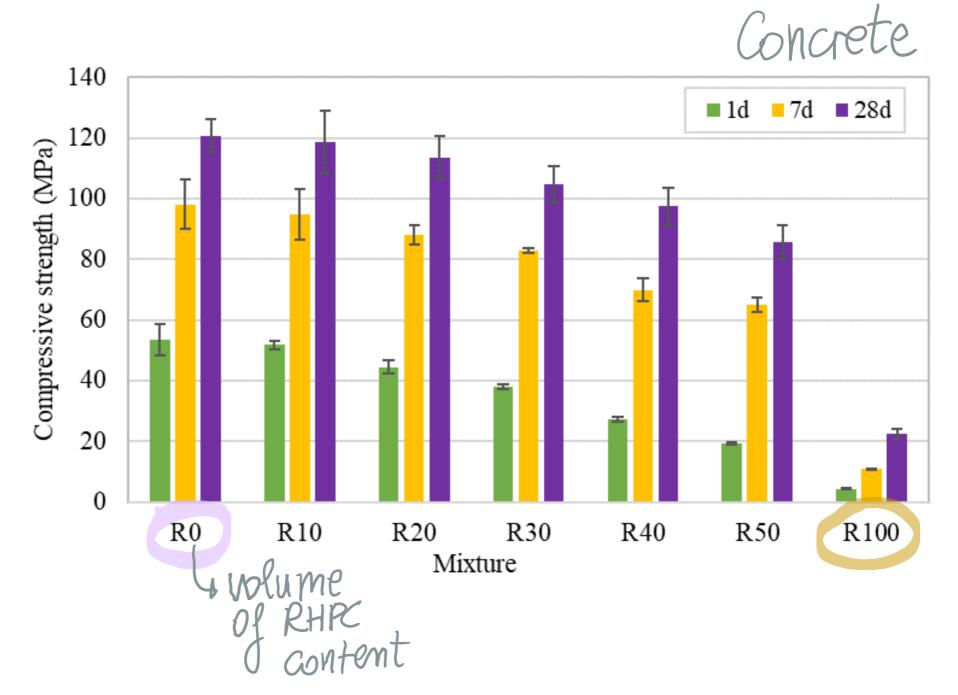




Phase name	OPC (wt%)	mRHPC (wt%)
C_2S	12.4 ± 1	2.9 ± 1
C_3S	66.1 ± 5	1.5 ± 1
C_4AF	12.6 ± 1	2.1 ± 1

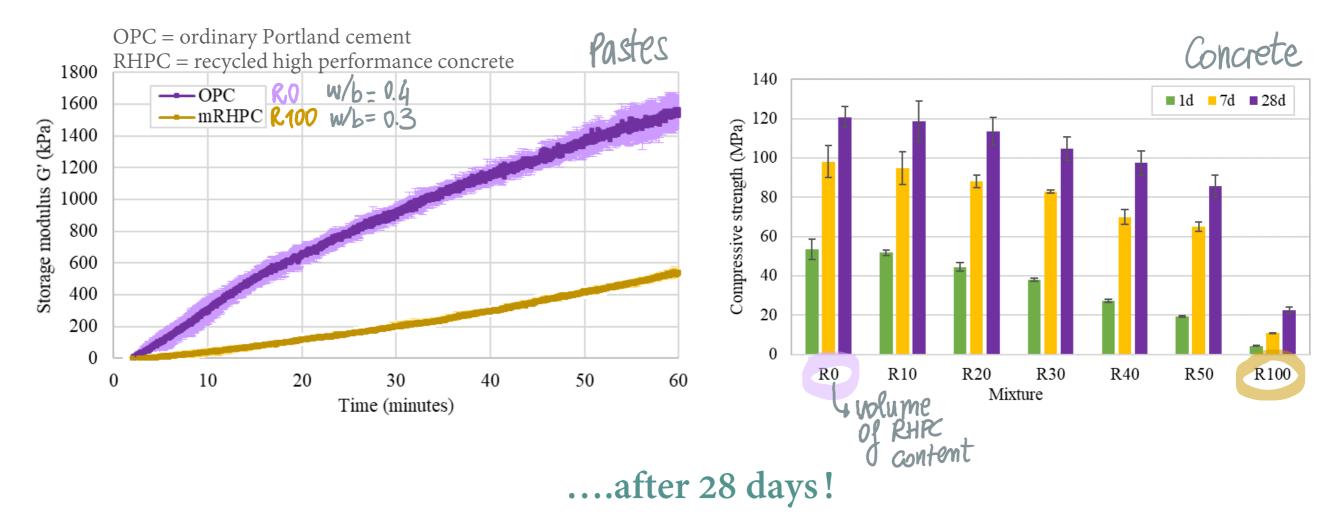








The residual reactivity detected by SAOS is confirmed by mechanical testing





Cement Chemistry Particle Interaction

Rheology SAOS

Sustainability

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