

17. Kolloquium und Workshop RHEOLOGISCHE  
MESSUNGEN AN MINERALISCHEN BAUSTOFFEN,  
12. UND 13.03.2008

# THE AIR-ENTRAINMENT METHODOLOGY DECREASE OF SELF-COMPACTING CONCRETE

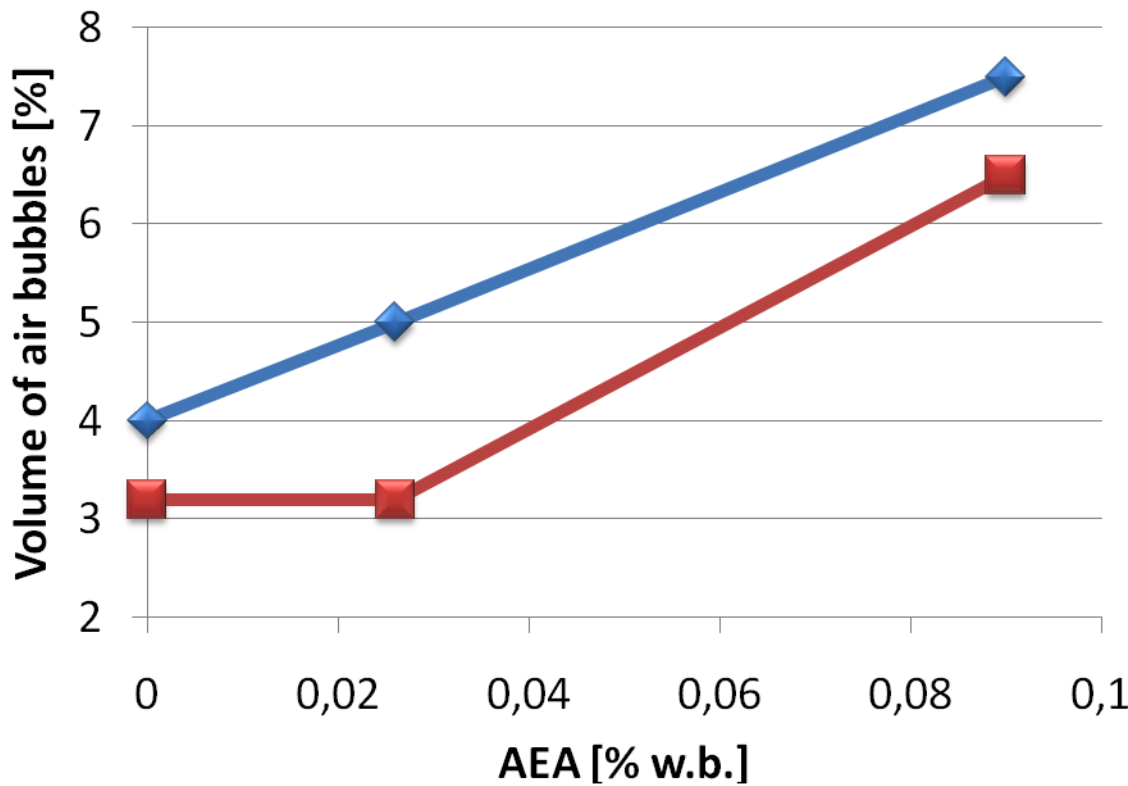


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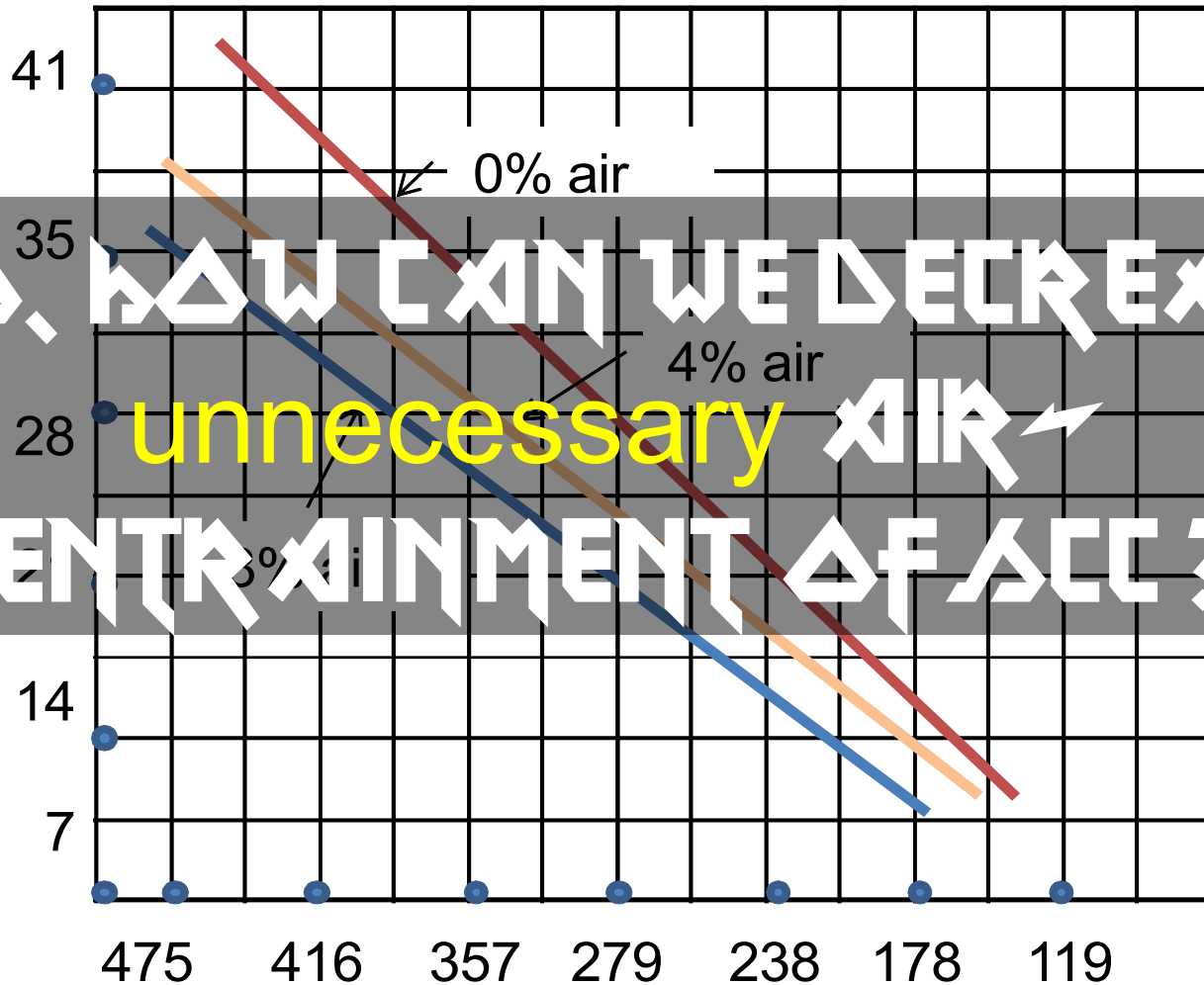
IT ACC WITH SP (PC,  
 EVEN IF THE AIR  
 NOT USED

As we can see  
 rheological

The flow SCC of 200 millimeters diameter included 3,2 % of air, and the flow of SCC with 720 millimeters diameter, included 4% of air.

In next step of laboratory research air-entraining admixture was inserted into the same SCC composition. It caused bigger air-entrainment of SCC in proportion to dosage of air-entraining admixture. So, and in this case rheological

# AIR VOLUME IN CONCRETE DECREASE OF CONCRETE COMPRESSIVE STRENGTH



SO, HOW CAN WE DECREASE

unnecessary AIR

ENTRAINMENT OF ACC?

Constant slump

# INFLUENCE OF RHEOLOGICAL PROPERTIES OF SCC ON AIR BUBBLES

TOO LOW VALUE OF

RHEOLOGICAL PARAMETERS  
PROPOSITION 1.

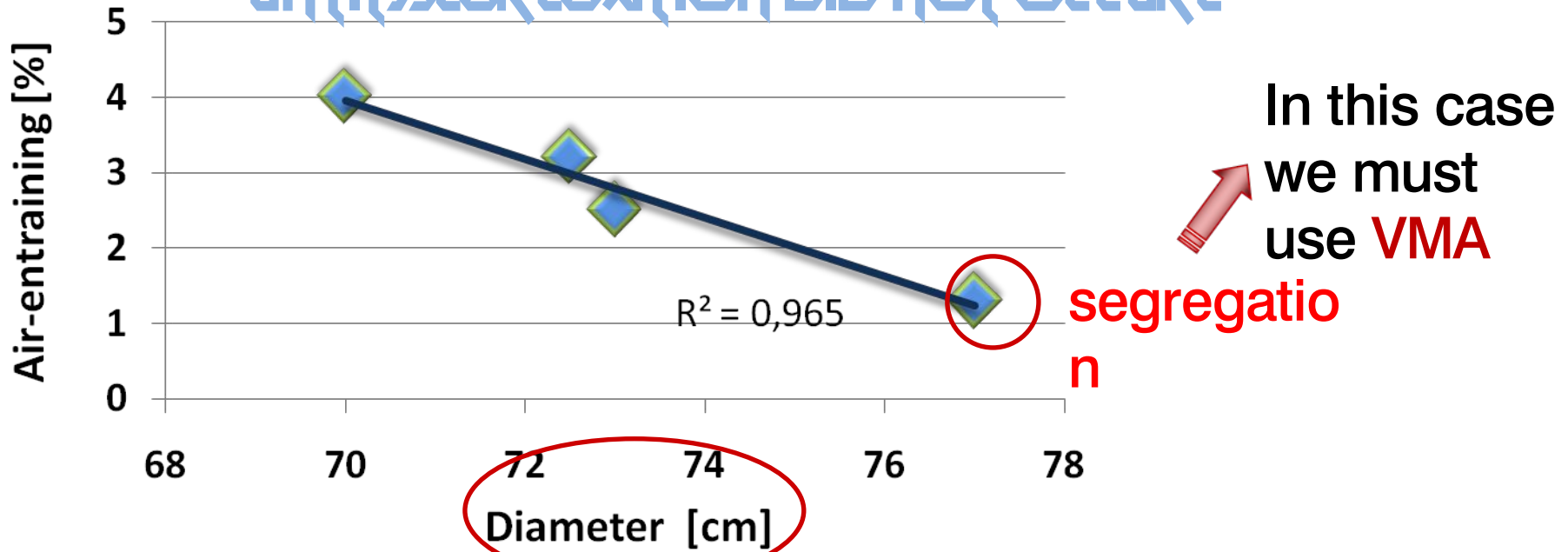
Adjusting

rheological

properties of SCC

DIMETER

LABORATORY RESEARCH SHOWED THAT SCC WAS NOT AIR-ENTRAINED WHEN DIAMETER FLOW WAS MAXIMIZED UNTIL SEGREGATION DID NOT OCCURE



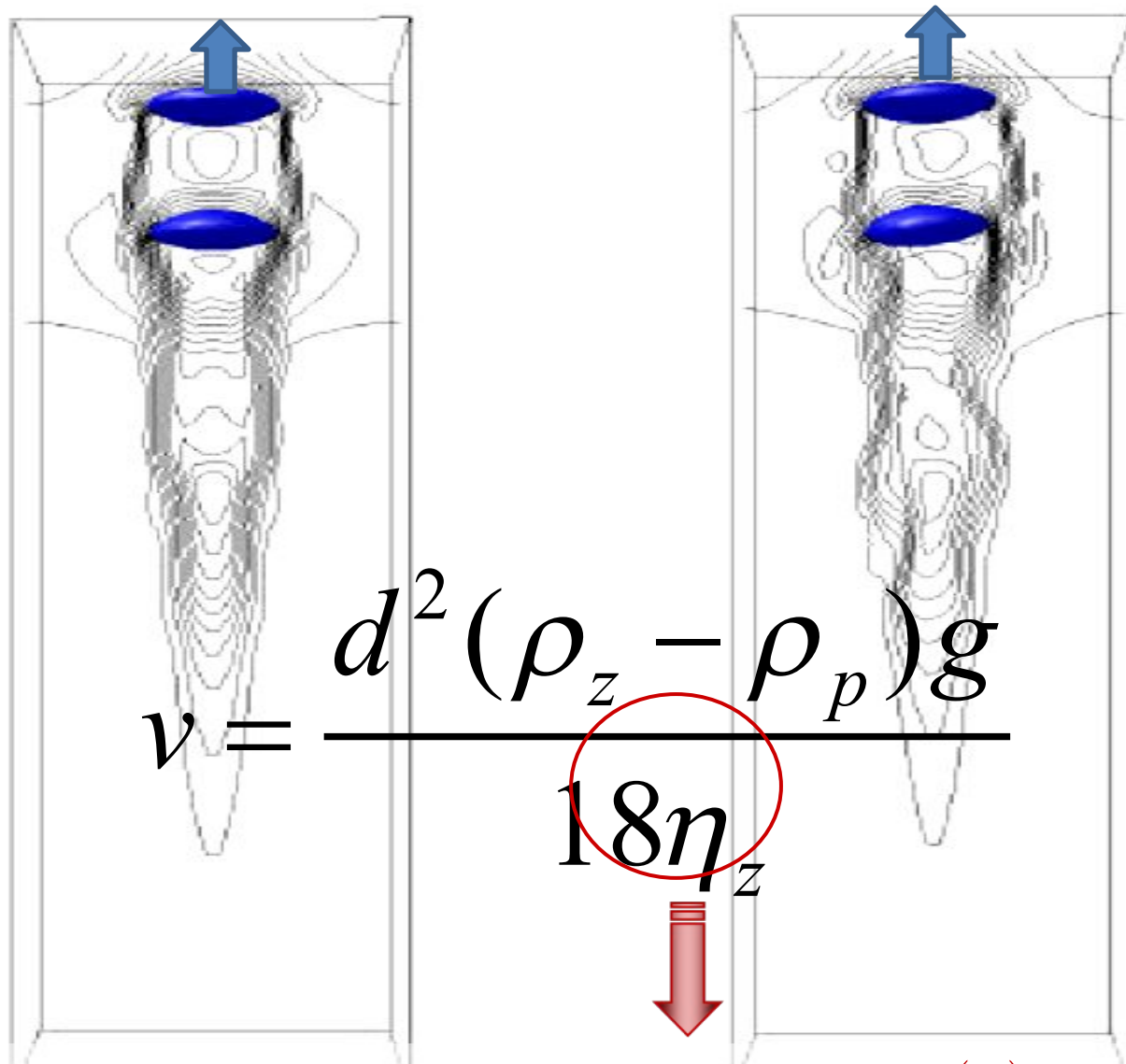
OUTFLOW QUALIFICATION OF AIR

$$(\rho_z - \rho_p) d \cdot g > \tau_0 \cdot A$$

diameter of air bubbles

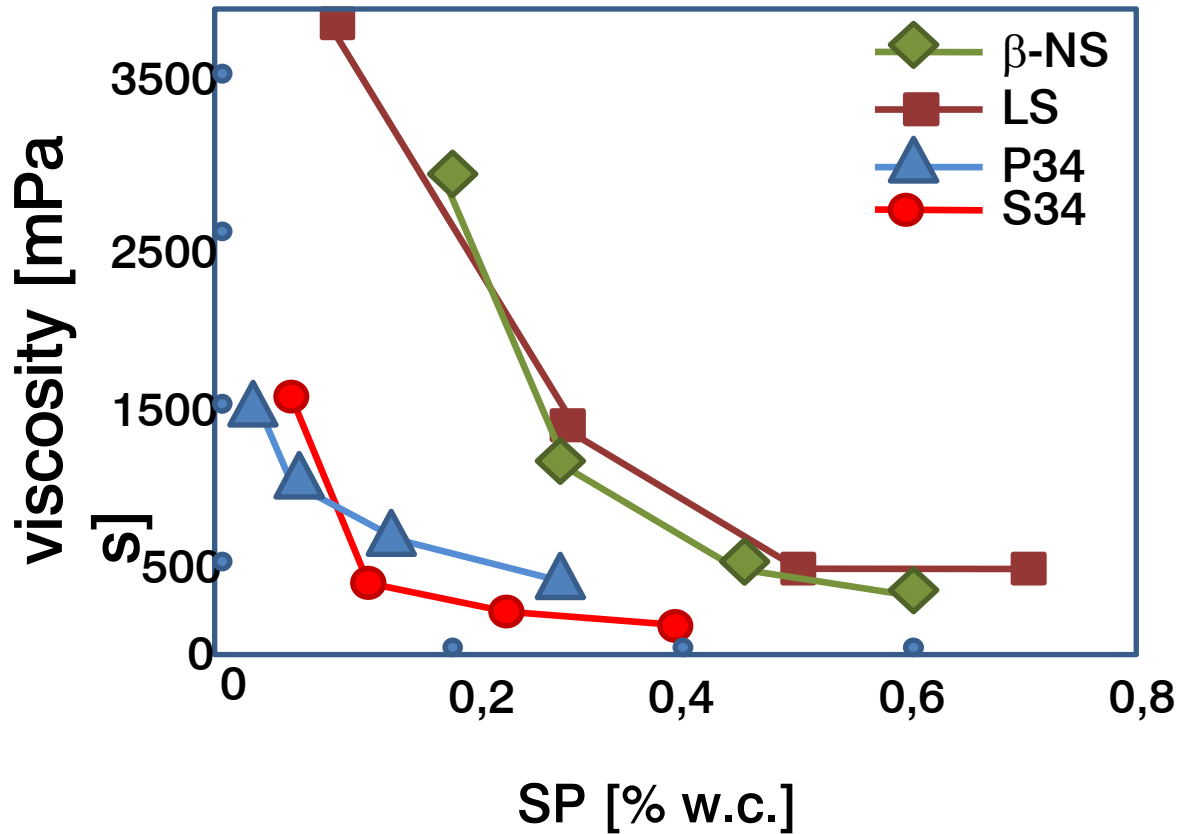
shape index of air bubbles

CONCLUSION: SCC WITH SP INCORPORATE NOT TOO MUCH AIR BUBBLES WHEN VALUE OF YIELD STRESS ( $\tau_0$ ) OF



**ADDITIONALLY, VALUE OF CEMENT PASTE VISCOSITY ( $\eta$ ) MUST BE LOW TO OBTAIN HIGH OUTFLOW SPEED OF AIR BUBBLE**

# INFLUENCE OF SP TYPE AND VALUE DOSE ON CONCRETE MIX VISCOSITY



IF WE KNOW THE VALUE OF CEMENT PASTE VISCOSITY AND YIELD STRESS WE CAN  
SHAPE THEM USING TYPE AND DOSE SUPERPLASTICIZER TO

# Superplasticizer type and its air-entraining effect

## Air-entraining effect of SP

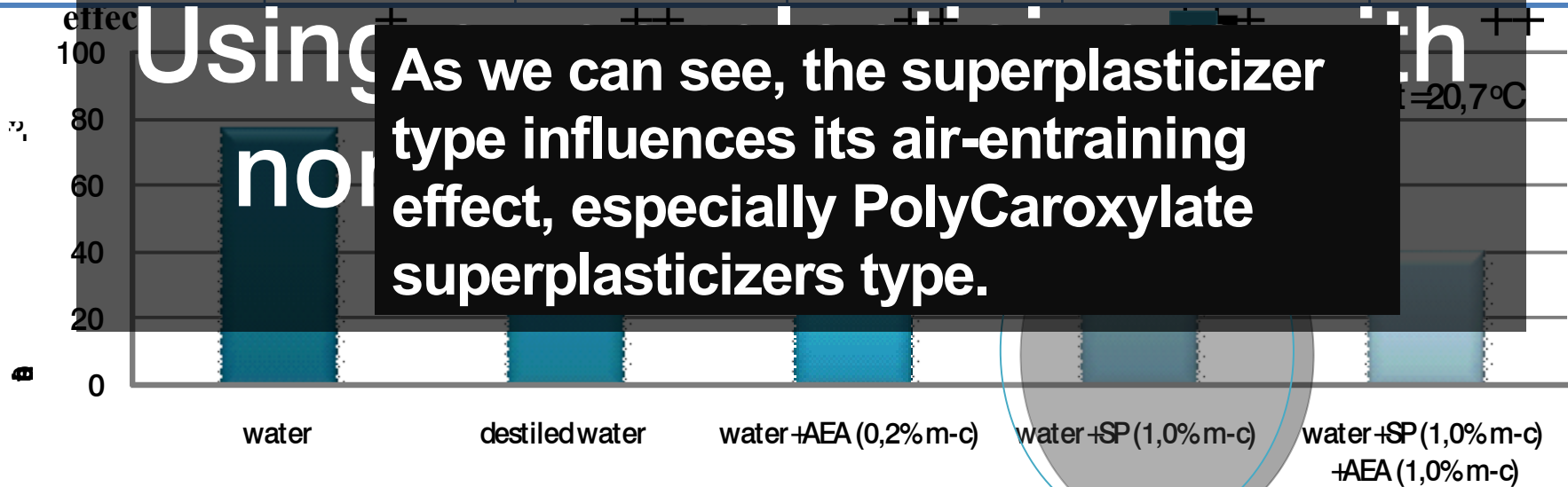
is caused i.a. by its influence on surface tension of water

### Proposition 2.

As we can see, the superplasticizer type influences its air-entraining effect, especially PolyCaroxylate superplasticizers type.

Air-entraining

effect



Superplasticizers decrease the value of surface tension as much as the air-entraining admixture.







The rate of surface tension decreases by usage surface active compounds is followed in accordance with **GIBBS EQUATION**:

$$-\frac{d\sigma}{dc} = RT \frac{\Gamma_i^s}{c}$$

where:

$d\sigma$  – change of surface tension

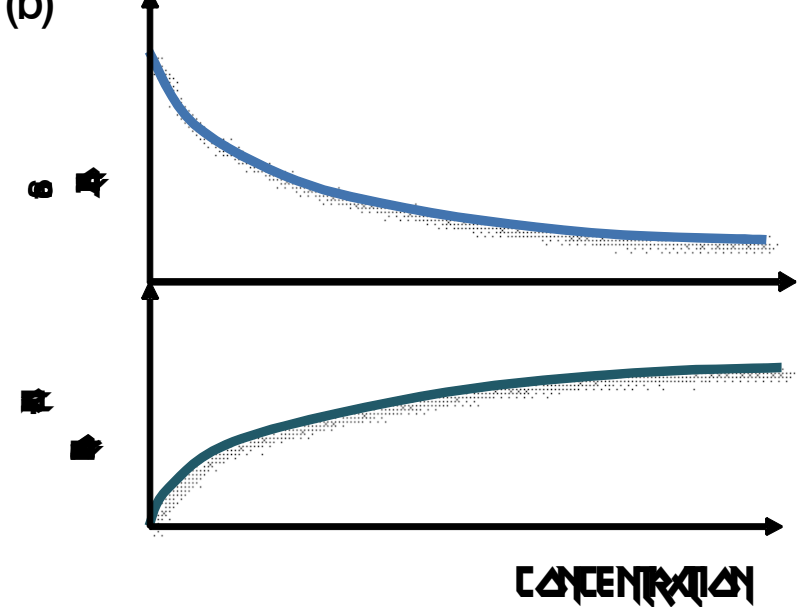
$dc$  – change of given substance concentration in the solution,

$\Gamma_i^s$  – surface concentration (mol/m<sup>2</sup>),

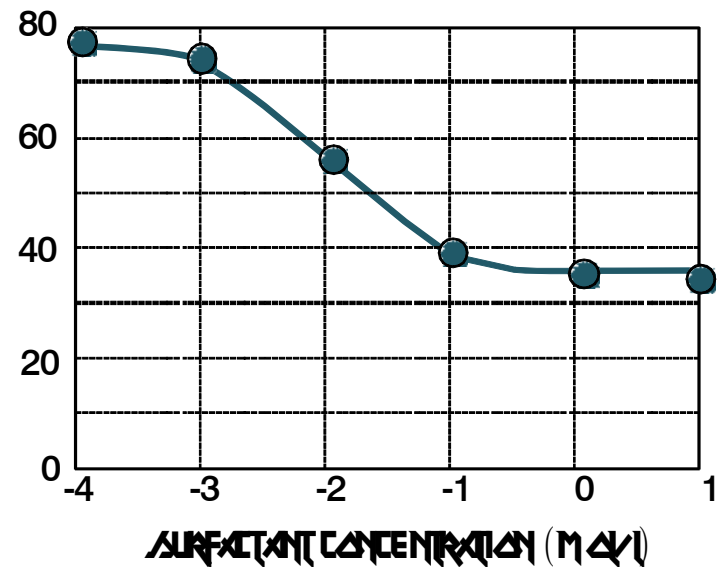
$c$  – constituent concentration,

$R$  – gas constant,

$T$  – temperature.

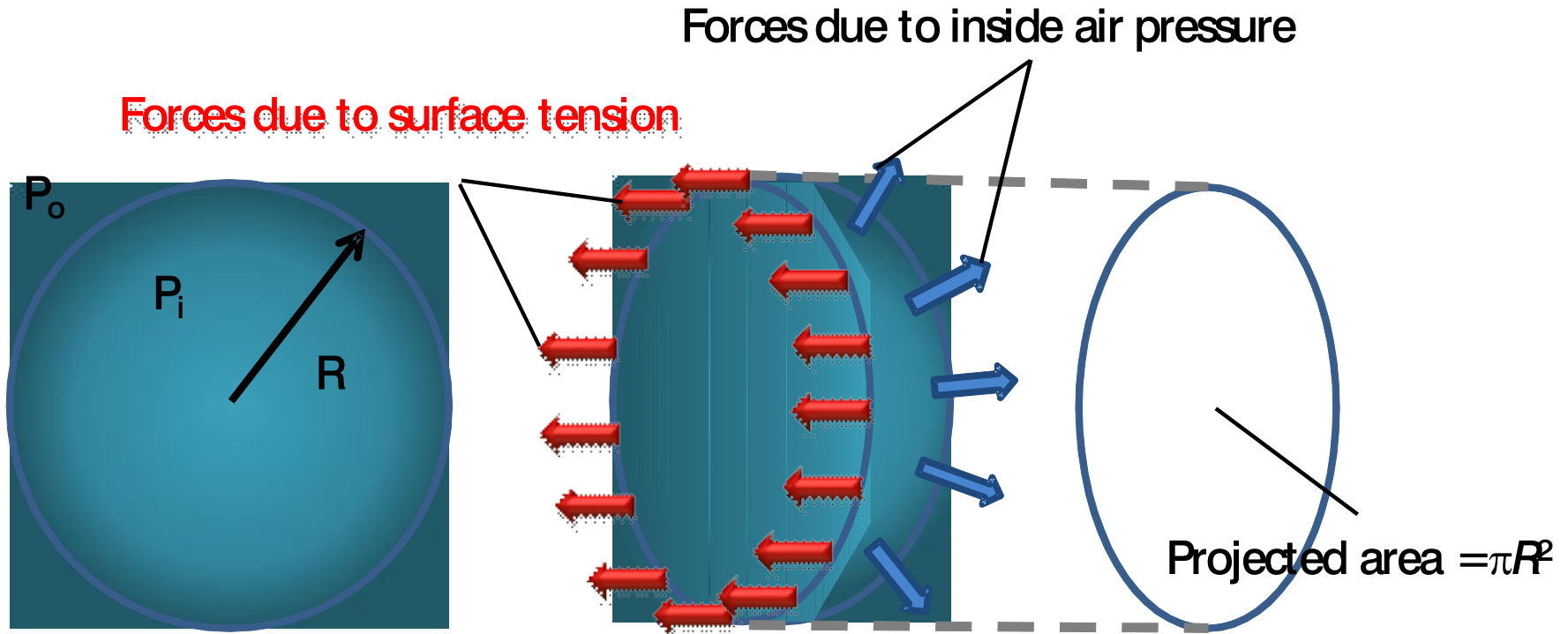


*Rapid change of surface tension and foam production over a concentration range*



*Rapid change of surface tension over a relatively small concentration range*

# IN THE PICTURE WE CAN SEE, HOW AIR BUBBLE IS FORMATED BY THE SURFACE TENSION



(a) Spherical air bubble

(b) free-body diagram for right-half of air bubble

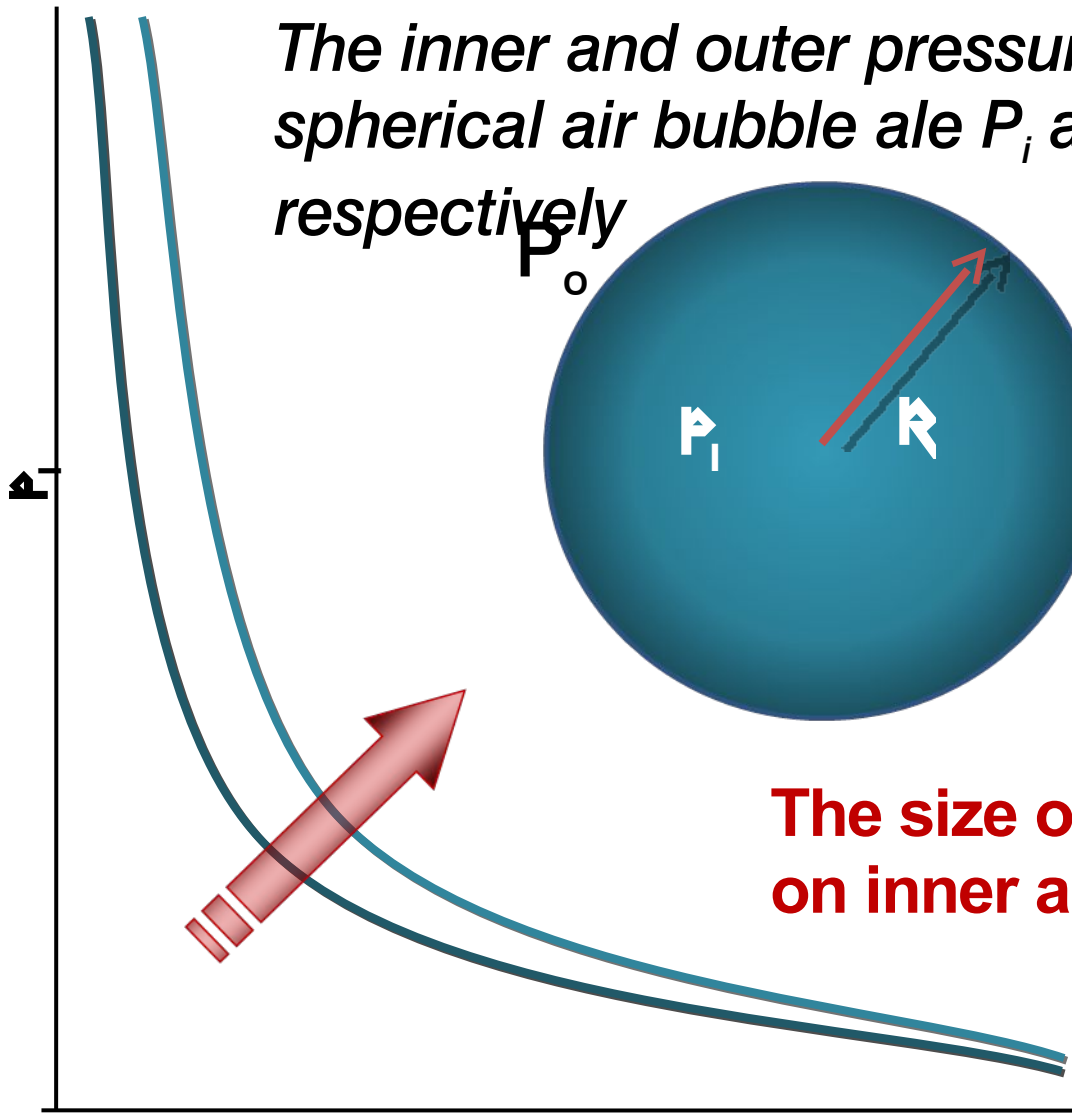
*the forces pointing to the left are due to the surface tension. The forces pointing perpendicular to the hemispherical surface are due the air pressure inside the bubble*



$$P_i - P_o = 2g / R \quad - \text{Laplace's law}$$

The inner and outer pressures on the spherical air bubble are  $P_i$  and  $P_o$ , respectively

PRESSURE INSIDE AN AIR BUBBLE,



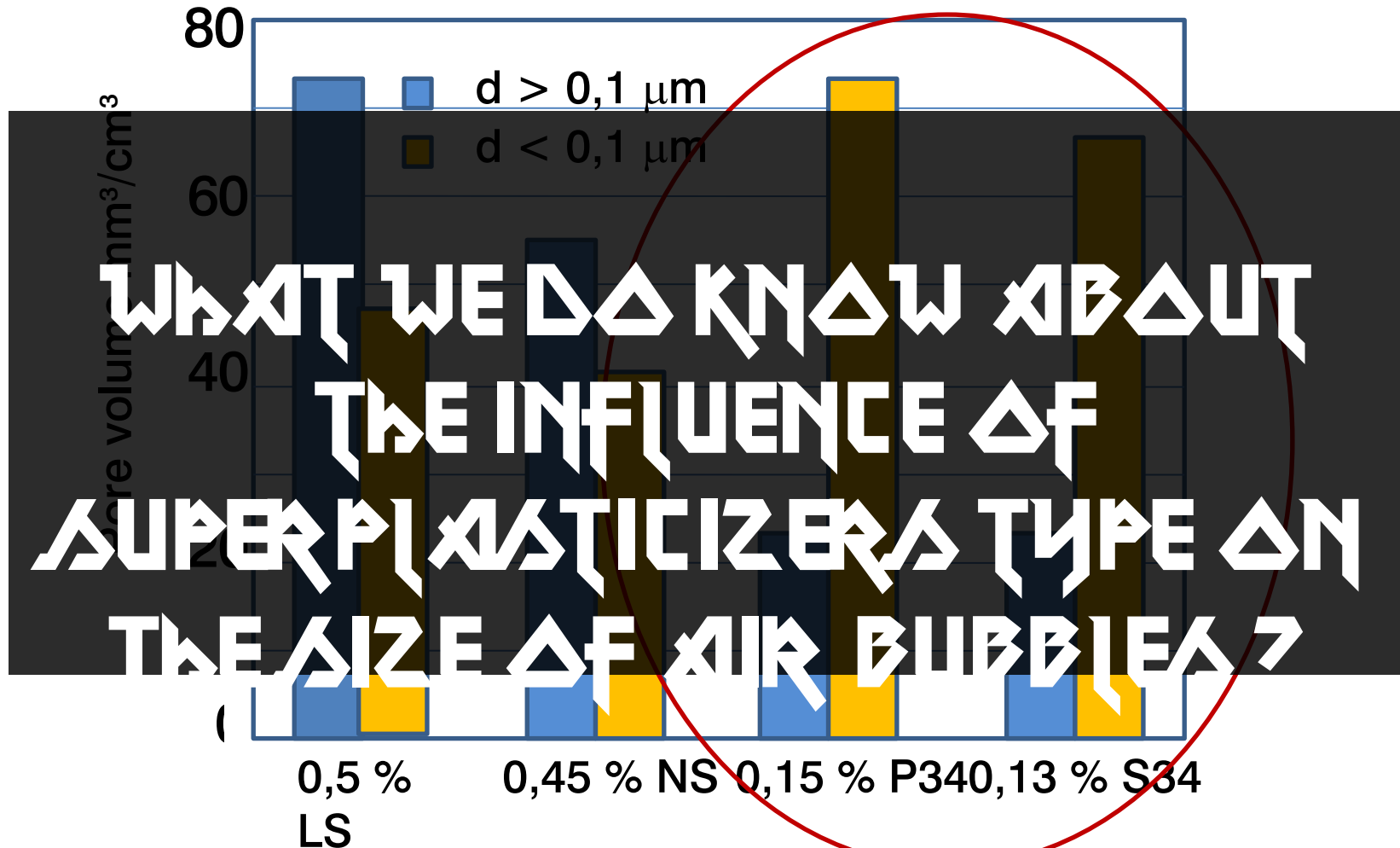
The size of air bubbles depends on inner and outer pressures

$P_o$

RADIUS, R

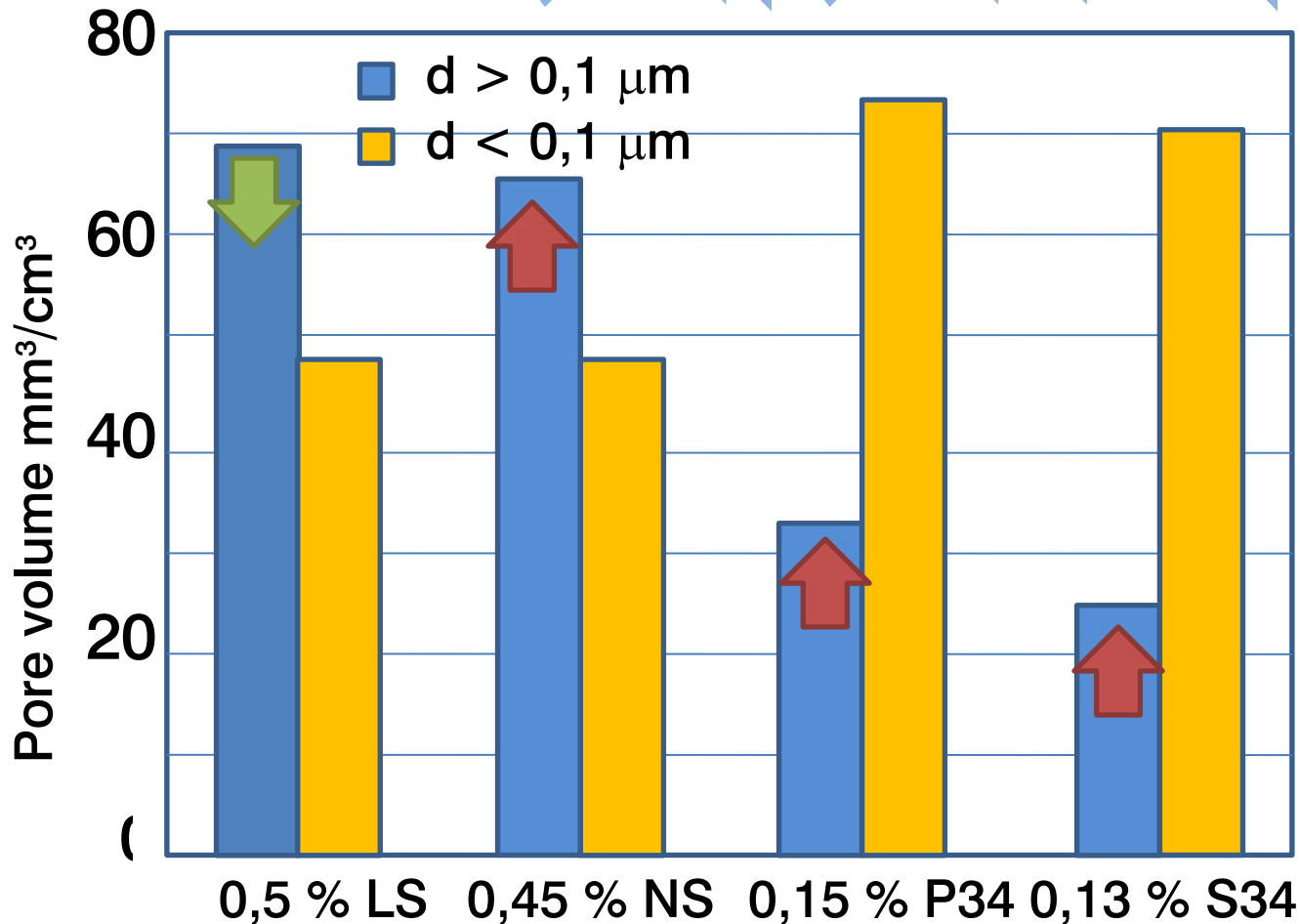
# INFLUENCE OF NAPHTHYLENE (NS), REFINED LIGNIN SULFONATE (LS) I POLYCARBOXYLATE (P34, S34) SUPERPLASTICIZER ON POROSITY STRUCTURE OF (28 DAYS)

PolyCarboxylate superplasticizers decrease the volume of pore diameter more than other superplasticizer

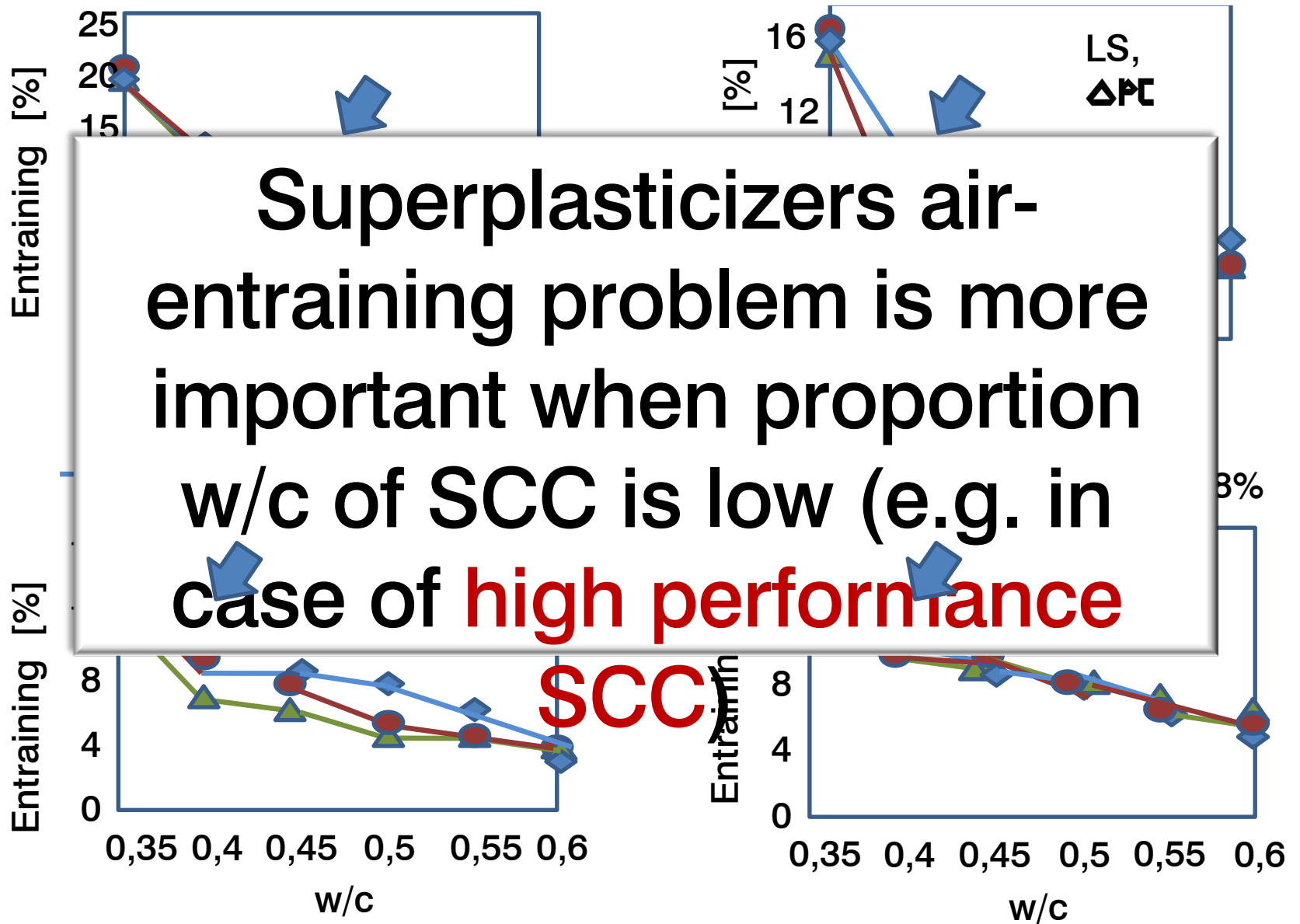


WHAT WE DO KNOW ABOUT THE INFLUENCE OF SUPERPLASTICIZERS TYPE ON THE SIZE OF AIR BUBBLES?

**BUT VOLUME OF PORE WITH  $d > 0,1 \mu\text{m}$  INCREASE AFTER 91 DAYS ...  $\Delta$ , ACCORDING THIS RESEARCH ALL TYPE OF SP CAUSE AIR-ENTRAINING OF SCC**



when the w/c proportion of SCC is low – air-entrainment of SCC is high



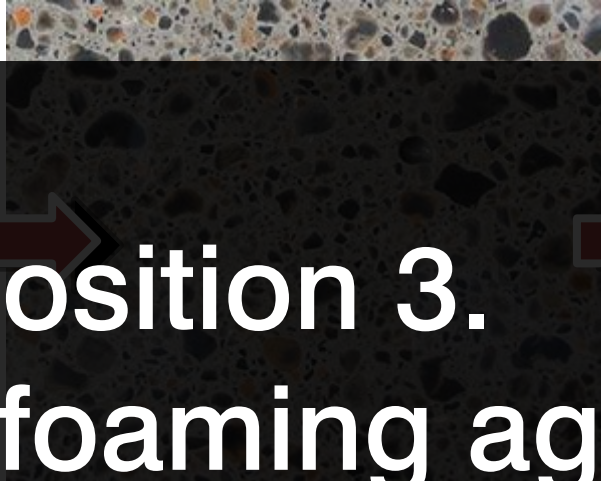
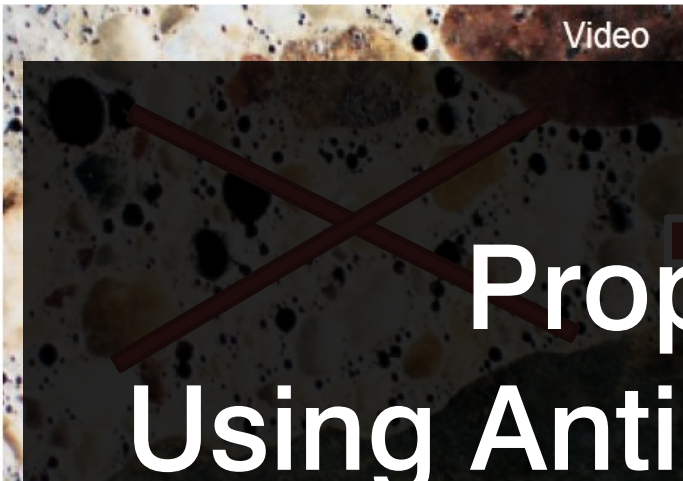
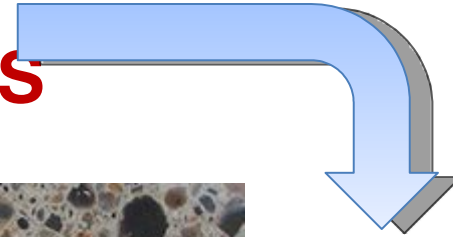
Lignosulfonic acid (LS), Portland cement (OPC), Low alkali cement (LAC), and white cement (WC)

# ANTI-FOAMING AGENTS:

e.g. Polyoxy Ethylene/propylene ether mixture

It causes:

can damage air bubbles



Proposition 3.

Using Anti-foaming agents

BETTER  
PERFORMANCE OF ACC

Some type of anti-foaming agents: ARE COMPATIBLE WITH POLYCARBOXYLATE TYPE SUPERPLASTICIZER AND MOST AIR-ENTRAINING AGENT, ESPECIALLY WITH VINYL TYPE AIR-ENTRAINING AGENT.





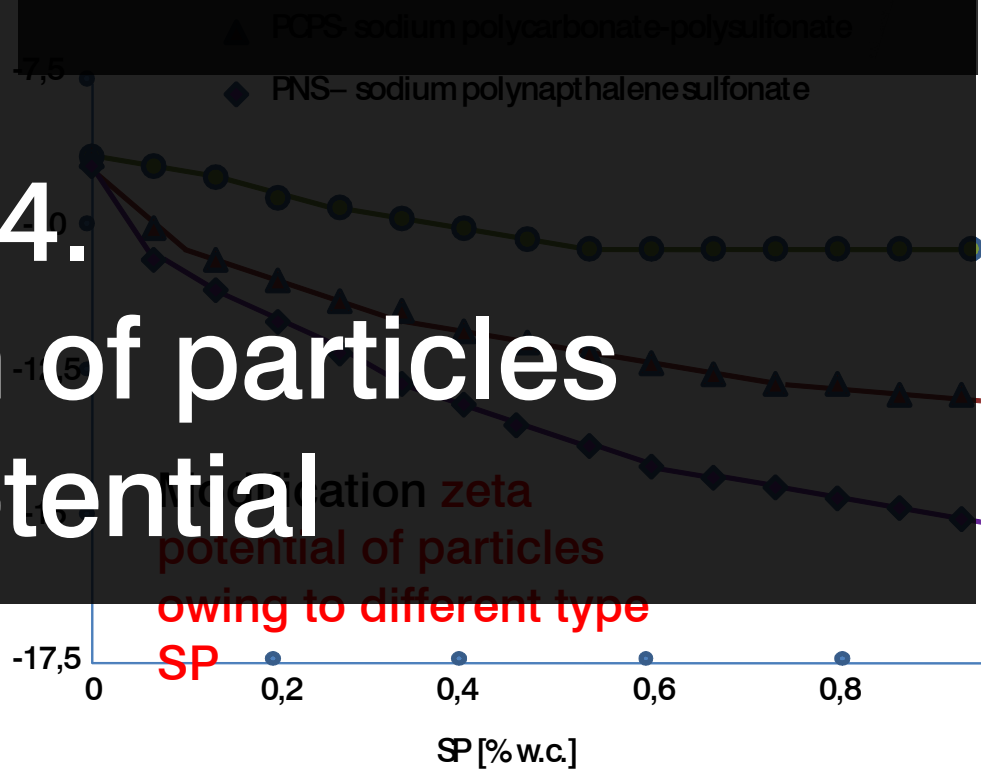
**DISPERSION RATE OF PARTICLES DEPENDS ON MODIFICATION EFFECT SURFACE LAYER OF PARTICLES BY TYPE SP**



Lon diffusion layer containing zeta potential  
**WE SHOULD USE THIS SP TYPE WHICH CAN CHANGE SURFACE LAYER OF PARTICLES TO NOT PERMIT AIR BUBBLES connect with particles**

Dispersion of air bubbles and cement particles and

**Proposition 4. Modification of particles electrical potential**



modification zeta potential of particles owing to different type SP



# CONCLUSIONS

HOW CAN WE DECREASE AIR ENTRAINMENT OF

1) ADJUSTING THE RHEOLOGICAL PROPERTIES OF SCC:

SCC WITH SP DO NOT INCORPORATE NOT TO MUCH AIR BUBBLES WHEN VALUE OF YIELD STRESS (  $\sigma$  ) OF VALUE OF VISCOSITY (  $\eta$  ) SHOULD NOT BE HIGH TO INCREASE

2) USING SUPERPLASTICIZERS WITH NON AIR ENTRAINING AFFECT

3) USING ANTI-FORMING AGENTS TO ELIMINATE AIR BUBBLES

4) MODIFICATION OF ELECTRIC PARTICLES POTENTIAL TO HELP AIR BUBBLES OUTFLOW

LABORATORY RESEARCH IS STILL CONTINUED AND ITS RESULTS WILL SHOW WHICH OF THIS METHOD IS MOST EFFECTIVE